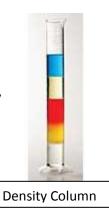
ICP: Density of Fluids

Objectives:

- 1. Learn how to find the density of liquids and use your understanding to make a density column.
- 2. Use a density column to predict the density of a solid.

Notes:

• A fluid is defined as any matter that is able to flow.

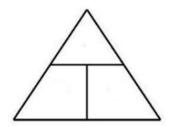


Density

1. What is density	y?
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Density is a _	of the amount of matter or	that can fit in a given
	. It is the ratio of a material's mass to its volum	e.

2. What is the formula for density?



3. What does density tell me about an object?

- o Density is important because it affects whether objects will _____ or ____.
- o If density is less than 1.0 g/mL, the object _____ in water.
- o If density is more than 1.0 g/mL, the object ______ in water.

4. Where would I see density in my everyday life?

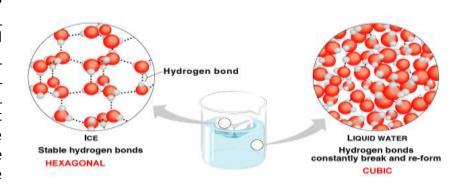
- ______ you want the balloon to get off the ground.
- * you don't want the ship to sink! People don't like to be on a sinking ship.
- *______ you don't want to overload the truck!
- * ______ you want to make sure you actually bought a gold necklace.

5. Density values chart

Object	Density Value (g/mL or g/cm³)	Sinks or Floats in water?	
Water (4°C)			
Ice water (0°C)			
Aluminum			
Silver (solid)			
Silver (liquid)			
Gold			
Iron			
Gasoline			
Baby Oil			

6.	REMEMBER : the phase of matter depends on it	ts The		the temperature,
	the farther the molecules are spread out and the	e faster they are moving.	Therefore a	of one
	material will be denser than a of the sa	ame material.		

An	excep	tion	to	the	rule	is
	·		Sol	id		
() is	less	den	se th	an liqi	uid
		Th	at is	s wh	У	
float	s in y	our	drink	ːs		
mole	ecules	free	eze	into		
cryst	als tha	at fo	rm a	patt	ern tl	nat
has	a lot	of e	mpty	, spac	ce. T	he
mole	cules i	n liq	uid	•	ā	are
	_					
	float mole cryst has mole	floats in your molecules crystals that has a lot molecules is more tight	(Sol () is less den That is floats in your drink molecules freeze crystals that form a has a lot of empty molecules in liquid _	Solid () is less dense the That is wh floats in your drinks molecules freeze into crystals that form a patt has a lot of empty space molecules in liquid more tightly packed.	() is less dense than lique



8. Practice Problems

9. Clay Practice: A review of triple-beam balances and displacement method

Clay Mass (g)	Volume (mL)	Density (g/mL)	Density (g/mL) (rounded to nearest whole number)	Sinks or Floats?
10				
20				
30				