

Type I and II Ionic Compounds

Directions:

1. Write your first and last name on your paper
2. Write your Chemistry period number under your name
3. Number your paper 1 to 10.
4. Just write the answers to each problem on your paper.

Type I. Write the compound name of the chemical formulas below.

1. $\text{MgBr}_2 =$ _____
2. $\text{CaO} =$ _____
3. $\text{Li}_2\text{S} =$ _____
4. $\text{KI} =$ _____
5. $\text{Be}_3\text{P}_2 =$ _____

Type 2. Write the chemical formulas for the compounds below.

- 6) vanadium (V) sulfide _____
- 7) chromium (III) oxide _____
- 8) platinum (IV) iodide _____
- 9) lead (II) nitride _____
- 10) gold (III) bromide _____