$\qquad$
$V$ is in milliliters ( mL ) or cubic centimeters $\left(\mathrm{cm}^{3}\right)$ M is in grams ( g )
$D$ is in $\mathrm{g} / \mathrm{mL}$ or $\mathrm{g} / \mathrm{cm}^{3}$

1.) Find the density of a substance whose volume is 5 mL and whose mass is 25 g .
2.) Calculate the mass of a substance whose density is $0.2 \mathrm{~g} / \mathrm{mL}$ and whose volume is 45 mL .
3.) Find the volume of a substance whose density is $4 \mathrm{~g} / \mathrm{mL}$ and mass is 4.1 g .
4.) Find the mass of a substance whose density is $8.1 \mathrm{~g} / \mathrm{mL}$ and volume is 2.95 mL .
5.) Find the volume of a substance whose density is $1.93 \mathrm{~g} / \mathrm{mL}$ and mass is 31.3 g .
6.) Find the density of a substance whose volume is 45.2 mL and mass is 5 g .
7.) Calculate the volume of a substance whose density is $0.90 \mathrm{~g} / \mathrm{mL}$ and mass is 7.1 g .
8.) Calculate the mass of a substance whose volume is 3.15 mL and density is $0.79 \mathrm{~g} / \mathrm{mL}$.
9.) If a substance has a volume of 5.1 mL and a mass of 2.7 g , what is its density?
10.) If a substance has a volume of 10 mL and its density is $1.26 \mathrm{~g} / \mathrm{mL}$, what is its mass?

