Ms. Young

Science _____

October 2016

Law of Conservation of Mass

Objectives: To observe the Law of Conservation of Mass

Materials: 1 balloon, weighing paper, 125-mL Erlenmeyer flask, scoopula, TBB, graduated cylinder, sodium bicarbonate, acetic acid, funnel

Procedure Part A:

- 1. Record the mass of the empty Erlenmeyer flask
- 2. Record the mass of the empty balloon
- 3. Measure 15 mL of acetic acid
- 4. Pour into the Erlenmeyer flask
- 5. Mass the flask and acetic acid. Record.
- 6. Calculate the mass of the acetic acid ONLY. Record in your data table. Hint: Use subtraction!
- 7. Mass 5g of sodium bicarbonate
- 8. Pour the sodium bicarbonate into the balloon. NOTE: You will need to problem solve on this step!
- 9. While one student holds the flask, another must slip the open end of the balloon over the mouth of the flask, while keeping the sodium bicarbonate from entering the flask.
- 10. Add the masses in the PART A data table.

Procedure Part B:

- 11. Tip the balloon upright, allowing the sodium bicarbonate to drop into the flask and allow the reaction to fully complete. Swirl your reaction a little bit to make sure all of the sodium bicarbonate reacted.
- 12. After the reaction is completed, mass the system (balloon, flask, and products). Do not let any gas escape from the system while you measure the mass!
- 13. Record your findings.

Procedure Part C:

- 14. Open the balloon to let the gas escape.
- 15. Measure the mass of the balloon, flask, and products.
- 16. Record
- 15. Follow clean up instructions posted on the board!



Data for PART A:

Material	Mass (g)
Erlenmeyer Flask (empty)	
Balloon (empty)	
Acetic acid	
(mass of acid ONLY)	
Sodium bicarbonate	
(mass of sodium bicarbonate ONLY – not the weighing paper)	
Total Mass of all objects	

Data Procedure B

Material	Mass (g)
Mass of entire system AFTER reaction stops	

If everything went PERFECT during your lab, you will notice something about these two numbers!

Data	Procedure C			This number
	Material	Mass (g)		should be
Μ	lass of entire system AFTER you released the gas from the balloon.			slightly lower than your mass
			-	in part A!



